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SAEEE 2025 Sample Question Paper

Sathyabama All India Entrance Examination

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#### MATHEMATICS

1. The quadratic equation x<sup>2</sup>-6x+1=0 and x<sup>2</sup>-cx+6=0 have one root in common. The other roots of the first and second equations are integers in the ratio 4:3 then the common root is



- (B) Y
- (C) N
- (D) Y-X



- 5. The area of the region described by A={(x,y):  $x^2 + y^2 \le 1$  and  $y^2 \le 1-x$ } is:
- (A)  $\frac{\pi}{2} \frac{2}{3}$ (B)  $\frac{\pi}{2} + \frac{2}{3}$ (C)  $\frac{\pi}{2} + \frac{4}{3}$ (D)  $\frac{\pi}{2} - \frac{4}{3}$

### PHYSICS

- 1. An object is immersed in a fluid. In order that the object becomes invisible, it should
  - (A) behave as perfect reflector
  - (B) have refractive index one
  - (C) absorb all light falling on it
  - (D) have refractive index matching with that of the surrounding liquid

2. If the rms velocity of the hydrogen molecules at NTP is 1.84 km/s, calculate the rms velocity of the oxygen molecules at NTP. Molecular weight of hydrogen and oxygen are 2 and 32 respectively.

- (A) 1.47 km/s
- (B) 0.94 km/s
- (C) 1.84 km/s
- (D) 0.47 km/s

3. Using an AC voltmeter, the potential difference in the electrical line in a house is read to be 234V. If the line frequency is 50Hz, the equation of the line voltage is

- (A) 220 Sin 100  $\pi$ t
- (B) 165 Sin 100 πt
- (C) 440 Sin 100  $\pi$ t
- (D) 331 Sin 100 πt



4. The fact that light of transverse wave derive its evidence by the support from the observation that

- (A) light wave undergo reflection
- (B) light can be diffracted
- (C) light travels in waves
- (D) light shows polarizing effects

5. Refractive index of material is equal to tangent of polarizing angle. It is called

- (A) Brewster's law
- (B) Lambert's law
- (C) Malu's law
- (D) Bragg's law

#### CHEMISTRY

1. In the standardization of Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> using K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub> by iodometry, the equivalent weight of K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub> is

- (A) Molecular weight / 2
- (B) Molecular weight / 6
- (C) Molecular weight / 3
- (D) Same as molecular weight

2. What product are expected from the disproportionation reaction of hypochlorous acid ?

- (A) HClO<sub>3</sub> and Cl<sub>2</sub>O
- (B) HClO<sub>2</sub> and HClO<sub>4</sub>
- (C) HCl and Cl<sub>2</sub>O
- (D) HCl and HClO<sub>3</sub>

3. Native silver metal forms a water soluble complex with a dilute aqueous solution of NaCN in presence of

- (A) Nitrogen
- (B) Oxygen
- (C) Carbon dioxide
- (D) Argon

4. The number and types of bonds between two carbon atoms in calcium carbide are

- (A) One sigma, one pi
- (B) One sigma, two pi
- (C) Two sigma, one pi
- (D) Two sigma, two pi



- 5. Identify the incorrect statement among the following
  - (A) Ozone reacts with SO<sub>2</sub> to give SO<sub>3</sub>
  - (B) Silicon reacts with NaOH(aq) in the presence of air to give  $Na_2SiO_3$  and  $H_2O$
  - (C)  $Cl_2$  reacts with excess of  $NH_3$  to give  $N_2$  and  $NH_4Cl$
  - (D)  $Br_2$  reacts with hot and strong NaOH solution to give NaBr, NaBrO<sub>4</sub> and H<sub>2</sub>O